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Simultaneous fractionation of sulfur dioxide explains mass independent fractionation of sulfur isotopes in Archean sedimentary pyrites

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ABSTRACT

The relationship between $\Delta^{36}S$ and $\Delta^{33}S$ in Archean sedimentary pyrites has been used to evaluate early geologic processes, including photochemical reactions in the anoxic atmosphere, biological activity and thermochemical alteration during sediment deposition. We have applied statistical methods to quadruple S isotope analyses of Archean sedimentary pyrites, using data compiled from the literature. Most of the best-fit lines, on plots of Δ^{36} S against Δ^{33} S, have Archean reference array-like Δ^{36} S/ Δ^{33} S slopes that vary between -1.5 and - 0.9. Rigorous statistical tests were conducted to calculate the probability of the best-fit lines passing through the origin. Seventeen of 23 Δ^{36} S- Δ^{33} S regression lines, which pass our reliability filter of $R^2 \ge 75\%$ and Δ^{33} S range $\ge 2\%$, have positive intercepts on the Δ^{36} S axis, and 13 of these have a probability of < 5% of a zero intercept on the Δ^{36} S axis. The observed Δ^{36} S/ Δ^{33} S slopes and the non-negative intercepts, which requires at least two massindependent fractionation source reactions to operate simultaneously, can be produced by UV radiation in the atmosphere at low SO₂ partial pressures by combining collision-induced intersystem crossing in the SO₂ photoexcitation band (240-340 nm), with the self-shielding effect in the SO₂ photolysis band (190-220 nm). The two SO₂ photochemical processes must occur simultaneously in a single atmospheric reservoir in order that the fraction contributed by the end-member process remains constant across the full range of $\Delta^{33}S$ values. We call this process simultaneous fractionation. We applied a two-end-member model to calculate the fraction of S contributed by the SO₂ photoexcitation end-member (f) needed to produce the observed $\Delta^{36}S/\Delta^{33}S$ gradients and variable intercepts on the $\Delta^{36}S$ axis in the Archean sedimentary pyrites, when the other end-member is SO₂ photolysis with the self-shielding. The simplest explanation for variations in f, and therefore variations in Δ^{36} S/ Δ^{36} S gradients, is that it is controlled by changes in the partial pressure of SO₂ in the atmosphere.

1. Introduction

Mass independent fractionation (MIF) of 33 S, measured as Δ^{33} S, is the departure in 33 S from the expected relationship between δ^{33} S and δ^{34} S predicted from mass dependent fractionation (Farguhar et al., 2000) and it occurs in an oxygen free atmosphere when volcanic SO_2 dissociates into S^0 and SO_4^{2-} in the presence of ultraviolet light (Farquhar et al., 2001). Ono et al. (2003) suggested that the Archean atmospheric photochemical reactions produced S⁰ with positive Δ^{33} S and δ^{34} S, and H₂SO₄, that is negative in both isotopes. MIF of ³³S is accompanied by MIF of ³⁶S, measured as Δ^{36} S, so that when Δ^{36} S is plotted against Δ^{33} S the data lie on a fractionation line. The slope of this line depends on the wavelength of the ultraviolet light and composition of the atmosphere in which SO₂ photochemistry takes place (Masterson et al., 2011; Ono et al., 2013; Whitehill et al., 2015; Whitehill and Ono, 2012; Whitehill et al., 2013). If, as is generally assumed, the isotopic composition of sulfur in Earth's mantle is chondritic, mantle Δ^{36} S, Δ^{33} S, and δ^{34} S are near zero (Labidi et al., 2014). If Δ^{36} S is fractionated from Δ^{33} S by a single MIF process the fractionation line must therefore pass through the origin.

Archean sedimentary pyrites typically show a correlated variation between Δ^{36} S and Δ^{33} S values with Δ^{36} S/ Δ^{33} S ratios between -1.5 and - 0.9 (Farquhar et al., 2000; Kaufman et al., 2007; Zerkle et al., 2012). The array formed by the correlation between Δ^{36} S and Δ^{33} S, together with the $\Delta^{33}S\text{-}\delta^{34}S$ relationship with a slope of ${\sim}0.9,$ have been termed the Archean Reference Array (ARA) (Ono, 2017; Ono et al., 2009).

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Measurements of four stable sulfur isotopes in pyrites from early Archean bedded barite deposits revealed significant deviations from the ARA-like $\Delta^{36}S/\Delta^{33}S$ array with correlation slopes lower than the ARA and negative intercept values on the Δ^{36} S axis (Muller et al., 2017; Roerdink et al., 2016; Shen et al., 2009). The origin of non-zero intercepts is uncertain. Shen et al. (2009) proposed microbial sulfate reduction (MSR) for the sedimentary barite from the \sim 3.47 Ga Dresser Formation, NW Australia, whereas a complex three-step mixing model has been suggested for hydrothermal barite deposits for both the 3.55-3.51 Ga Theespruit Formation, Lower Onverwacht Group in the Barberton Greenstone Belt, Swaziland (Roerdink et al., 2016) and the 3.2 Ga Sargur Group, India (Muller et al., 2017). Roerdink et al. (2016) interpreted the observed deviation from the ARA-like $\Delta^{36}S/\Delta^{33}S$ array and the negative Δ^{36} S intercept as the result of a three-stage process: 1) production of a precursor sulfide with a non-ARA $\Delta^{36}S/\overline{\Delta^{33}S}$ slope via dissociation of SO_2 by the self-shielding mechanism of Ono et al. (2013), 2) reworking of atmospheric sulfide into the hydrothermal system, and 3) mixing with the MSR-reduced seawater sulfate (Roerdink et al., 2016) or thermally reduced barite sulfate at higher temperature (Muller et al., 2017). In contrast, Archean sedimentary pyrites with ARA-like Δ^{36} S- Δ^{33} S linear trends show positive Δ^{36} S intercepts (Farguhar et al., 2013; Galić et al., 2017; Izon et al., 2017; Ono et al., 2009; Williford et al., 2016). Ono et al. (2009) fitted a least-square linear regression to the results of 13 analyses of pyrites from a sedimentary layer in the \sim 2.5 Ga Klein Naute Formation and obtained an equation of $\Delta^{36}S = -0.86^*\Delta^{33}S$ + 0.27. The Δ^{36} S/ Δ^{33} S slope of -0.86 was recognized as being ARA-like but the implications of the positive Δ^{36} S intercept was not addressed. Williford et al. (2016) also noticed a positive $\Delta^{3\hat{6}}$ S intercept of +1.4‰ for the best-fit line for Δ^{36} S- Δ^{33} S data for pyrites from the lower units of the ${\sim}2.5$ Ga Mt. McRae Shale and attributed it to the microbial sulfate reduction. MSR, however, shifts the $\Delta^{36}S$ to more negative values (Farquhar et al., 2003; Johnston et al., 2007; Ono et al., 2006b), making this interpretation highly unlikely. The origin of the positive Δ^{36} S intercepts, recorded by Archean sedimentary pyrites, with $\Delta^{36}S/\Delta^{33}S$ slopes similar to the ARA, has not been addressed. A notable exception is a study by Gallagher et al. (2017) who attributed a positive intercept in sedimentary pyrites from the ~2.5 Ga Gamohaan Formation, South Africa, to recycling of crustal S-MIF from local arc-volcanic eruptions.

In this study we apply statistical methods to the compiled pyrite quadruple S isotope data from global Archean sediments, and show that some of the fitted lines for the pyrite $\Delta^{36}S - \Delta^{33}S$ correlation arrays have statistically significant positive intercepts on the $\Delta^{36}S$ axis. We argue that the observed linear $\Delta^{36}S - \Delta^{33}S$ arrays cannot be the result of mixing of sulfur from two or more independent reservoirs but can be the product of the simultaneous fractionation of SO₂ in an anoxic atmosphere by a combination of photochemical reactions in the 250–330 nm and 190–220 nm bandwidths.

2. $\Delta^{36}S/\Delta^{33}S$ relationships of the Archean sedimentary pyrites

Linear regression lines were fitted to 42 suites of Archean sedimentary pyrites (Table A.1) for which quadruple sulfur isotope analyses are available. Least squares linear regressions for their Δ^{36} S- Δ^{33} S distributions, were solved using a Python code written specifically for this study (see Supplementary materials), and the results are listed in Table 1. The solutions were calculated by building a G matrix, and tested with Chisquare of goodness of fit, as well as Critical Chi-square for 5, 50, and 95%. The confidence and P-value for the probability of fit were also calculated. The external uncertainties for the compiled pyrite data, which were calculated by combining the single analysis uncertainties for each individual analysis measured with the standard deviation of the reference materials, were applied to the linear regression. The uncertainties in the parameters of the best-fit line were further amplified by taking the scatter of the data into account. This procedure leads to a conservative estimate of the *p*-value for the fitted line passing through the origin, and is therefore the most rigorous statistical test for a nonzero intercept.

Fig. 1 shows the linear regression lines for two Archean sedimentary pyrite suites, NF-LOW1 and NF-UP4A, from the ~2.6 Ga Nauga Formation (Farquhar et al., 2013; Izon et al., 2017). The NF-UP4A pyrite suite was measured by the SF₆ IR-MS method, whereas the NF-LOW1 suite was measured by in-situ SIMS technique. They both show excellent inverse correlations between Δ^{36} S and Δ^{33} S (Farquhar et al., 2013; Izon et al., 2017). The data for the two pyrite suites are distributed around linear correlation lines, as expected, and their slopes of $-1.21~\pm$ 0.01 and -1.04 ± 0.03 (1SD) fall within the ARA range (Fig. 1a and b, Table 1). What is unexpected is that the correlation lines fitted to the NF-UP4A and NF-LOW1 data do not pass within uncertainty of the origin, with intercepts of +0.49 \pm 0.11‰ and +0.81 \pm 0.03‰ (1SD) on the Δ^{36} S axis (Fig. 1c and d, Table 1). The calculated p values for the probability of the correlation lines passing through the origin are less than 0.000,002 and 0.000,001 respectively (Table 1, Fig. A.1). The Python code, used to calculate the *p*-values, takes into account both the scatter in unknown data, and the combined precision errors of unknowns and reference materials, which increases the calculated uncertainty on the intercept. Furthermore, the quadruple S isotopic compositions of the two sedimentary pyrite suites were obtained by two different analyzation techniques. Failure of the two Δ^{36} S- Δ^{33} S correlation lines to pass within uncertainty of the origin, therefore cannot be attributed to underestimation of analytical errors or to an instrumental effect

A literature survey has shown that failure of the NF-UP4A and NF-LOW1 Δ^{36} S- Δ^{33} S correlation lines to pass through the origin is not unique. We compiled Δ^{36} S- Δ^{33} S data from 38 Archean sedimentary suites with linear correlations ($R^2 \ge 50\%$). The reliability of slope and intercept of the Δ^{36} S- Δ^{33} S linear regression lines increases with the R² value and Δ^{33} S range. Therefore, we have considered only suites with $R^2 \ge 75\%$ and a $\Delta^{33}S$ range $\ge 2\%$ (Fig. 2). Twenty-three of the 38 Archean sedimentary suites passed these criteria. Seventeen of the 23 correlation lines have positive intercepts on the $\Delta^{36}S$ axis of up to +1.4% and 13 of these have a probability of less than 5% of the intercept on the Δ^{36} S axis being zero. Of the remaining 6 data sets, with negative intercepts on the Δ^{36} S axis, 2 have a moderate probability (> 20%) of passing through the origin. The other 4 have a probability of less than 1% of a zero intercept on the $\Delta^{36}S$ axis, the lowest being -1.07% \pm 0.22‰ (1SD). The probability of the intercepts on the Δ^{33} S axis being zero was also calculated and the results are similar to those reported for Δ^{36} S axis (Table A.2). Detailed descriptions of the linear fits for the compiled Archean sedimentary pyrites are given in Supplementary materials.

3. Non-zero intercepts for the Archean Δ^{36} S- Δ^{33} S correlations

The best-fit lines for the sedimentary pyrite suites NF-UP4A and NF-LOW1, together with the majority sedimentary pyrite suites from the other Archean locations (Farquhar et al., 2013; Farquhar et al., 2007; Izon et al., 2017; Kaufman et al., 2007; Ono et al., 2009; Thomazo et al., 2009; Thomazo et al., 2013; Williford et al., 2016; Zhelezinskaia et al., 2014), have statistically positive intercepts on both the Δ^{36} S and Δ^{33} S axes (Table A. 2; Fig. 2). The probability of the best-fit lines for the pyrite suites NF-UP4A and NF-LOW1 not having positive intercepts on the Δ^{36} S axis is low to negligible respectively, and this is also true for many of the other suites listed in Table 1 and plotted in Fig. A.1.

As noted in the Introduction, if primitive S, which is assumed to have $\Delta^{33}S = \Delta^{36}S = 0$, is fractionated by a single event, $\Delta^{36}S$ and $\Delta^{33}S$ will lie on a linear array that passes through the origin. The observation that the best-fit correlation lines for NF-UP4A and NF-LOW1, and samples from most other Archean locations, do not pass through the origin, requires that at least two processes are involved in the fractionation of $\Delta^{36}S$ from $\Delta^{33}S$. There are five fractionation processes that are known to produce variations in $\Delta^{33}S$ and $\Delta^{36}S$ (Fig. 3):

Table 1

Linear regression fitting results for the Archean sedimentary pyrites.

Age (Ma)	Stratigraphic units	Region	Sample ID	Method	No.	∆ ³³ S range (‰)	Δ^{36} S/ Δ^{33} S	σ	∆ ³⁶ S intercept	σ	R ² value	P-value*	Ref.
>3800 3481	Ujaraaluk metasediments Dresser Formation	Québec Western Australia	NGB	SF6	13	2.09	-0.85	0.18	-0.62	0.28	0.67	0.01	Thomassot et al., 2015
9401	Formation	Australia	DF1	SF6	15	1.29	-1.37	0.37	-0.59	0.25	0.52	0.008053	Shen et al., 2009
			DF2	SF6	22	4.61	-0.81	0.04	-0.21	0.05	0.93	0.000029	Ueno et al., 2008
			DF3	SIMS	23	1.92					no trend		Wacey et al., 201
	Mendon	South	210	onno	20	1.72					uonu		Busigny et al.,
334 259	Formation Mapepe Formation	Africa South Africa	MNF	SF6	5	2.47	-0.47	0.19	-0.65	0.29	0.67	0.010874	2017
0209			MF-										Roerdink et al.,
			BARITE	SIMS	81	1.04	-3.56	0.27	-0.70	0.07	0.69	0	2013 Roerdink et al.,
			MF-CHERT	SIMS	42	3.48	-0.87	0.11	0.08	0.11	0.51	0.237525	2013 Montinaro et al.,
			MF-SHALE	SF6	8	1.86	-0.64	0.13	-0.43	0.26	0.81	0.049095	2015
3220	Moodies Group Bababudan	South Africa	MDG	SF6	17	0.20					no trend		Saitoh et al., 202 Mishima et al.,
8137	Group	South India	BG	SF6	12	1.95	-1.47	0.21	-0.03	0.13	0.83	0.408052	2017
2919	Vanivilas Formation	South India	VF	SF6	7	1.77	-1.17	0.14	-0.13	0.09	0.94	0.081079	Mishima et al., 2017
2839	Mozaan group	South Africa	MG	SF6	8	0.88					no trend		Ono et al., 2006a 2006b
2740	Temagami greenstone belt	Canada	TGB	SF6	47	3.15	-0.74	0.07	0.042	0.06	0.74	0.226627	Diekrupa et al., 2018
			TGB-BIF TGB-	SF6	21	1.25	-1.38	0.21	0.351	0.14	0.70	0.005488	Diekrupa et al., 2018 Diekrupa et al.
			CHERT	SF6	8	0.55	-0.63	0.24	0.022	0.14	0.53	0.438438	Diekrupa et al., 2018
			TGB-VOLC	SF6	13	2.37	-0.57	0.08	0.029	0.08	0.81	0.351387	Diekrupa et al., 2018
			TGB-										Diekrupa et al.,
			SHALE	SF6	5	1.64	-0.90	0.08	0.342	0.13	0.98	0.003321	2018 Thomazo et al.,
720	Tumbiana Formation Kidd Creek	Western Australia	TF	SF6	34	1.89	-1.61	0.14	0.23	0.09	0.80	0.003172	2009; Farquhar et al., 2007
2713	rhyolitic complex	Canada	KCRC	SF6	62	4.90	-0.93	0.06	0.12	0.08	0.81	0.083647	Kurzweil et al., 2013
	Ingaldhal			0.5.4			- 						Mishima et al.,
2676	Formation Batatal	South India	IF	SF6 SF6 &	14	5.00	-0.77	0.11	-1.07	0.22	0.80	0.000001	2017 Zhelezinskaia
2660	Formation	Brazil	BF	SIMS	147	5.95	-0.93	0.04	0.20	0.06	0.79	0	et al., 2014
639	Jeerinah Formation	Western Australia	JF	SIMS	19	13.19	-0.89	0.04	0.89	0.45	0.97	0.023239	Williford et al., 2016
635	Lokammo Formation	South Africa	LF	SF6	9	4.90	-1.02	0.04	0.28	0.21	0.99	0.085229	Zerkle et al., 201
	Hiriyur	Couth India	LIE.	CE(10	F 0F	0.05	0.06	0.21	0.07	0.06	0.000001	Mishima et al.,
2633	Formation Carawine	South India Western	HF	SF6	13	5.05	-0.95	0.06	-0.31	0.07	0.96 no	0.000001	2017 Williford et al.,
2633	dolomite	Australia	CD	SIMS	15	0.78					trend		2016
2601	Cheshire Formation	Zimbabwe	CF	SF6	22	0.96	-1.66	0.24	0.26	0.15	0.71	0.043238	Thomazo et al., 2013
	Lower Nauga	South											
2588	Formation	Africa											Farquhar et al.,
			NF-LOW1	SIMS	156	10.88	-1.21	0.01	0.81	0.03	0.99	0	2013
	Upper Nauga	South	NF-LOW2	SF6	19	5.30	-0.98	0.09	-0.01	0.45	0.87	0.492908	Zerkle et al., 201
2549	Formation	Africa											Farquhar et al.,
			NF-UP1	SIMS	47	9.05	-0.97	0.05	-0.30	0.31	0.81	0.166639	2013
			NF-UP2	SF6	15	4.10	-1.07	0.26	-0.65	1.30	0.57	0.307455	Zerkle et al., 201
	Upper Nauga	South	NF-UP3	SF6	8	9.19	-0.90	0.01	-0.09	0.03	1.00	0.00031	Ono et al., 2009
	Formation	Africa											
552	FOIIIIAUOII												
2552	FOIIIIALIOII		NF-UP4H NF-UP4A	SF6 SF6	29 55	5.60 6.51	$-1.58 \\ -1.04$	0.12 0.03	1.14 0.49	0.54 0.11	0.86 0.96	0.016571 0.000002	Izon et al., 2017 Izon et al., 2017

Table 1 (continued) Sample ID Method $\Delta^{33}S$ $\Delta^{36}S/$ $\Delta^{36}S$ R^2 P-value* Age Stratigraphic Region No. Ref. σ σ $\Delta^{33}S$ (Ma) units range intercept value (‰) Gamohaan South Kaufman et al., Formation Africa 2007 Gallagher et al., Gamohaan South 2521 GF2 0.000557 SIMS 88 9.55 -1.000.02 -0.410.13 0.96 2017 Formation Africa Gallagher et al. GF2-STEEP SIMS 17 3.99 -1.190.12 0.32 0.31 0.88 0.151732 2017 GF2-Gallagher et al., SHALLOW 0.95 2017 71 5.62 -0.910.02 -0.990.14 0 SIMS Klein Naute South 2521 Africa KNF SF6 21 5.33 -1.010.08 0.25 0.14 0.90 0.033467 Ono et al., 2009 Formation Western Kaufman et al.. 2501 Mt. McRae Shale MMS1 SF6 42 11.52 -0.850.09 0.96 0.02213 2007 Australia 0.03 0.17 Kaufman et al. MMS1-UP 25 5.63 -1.020.30 0.11 0.93 0.004403 2007 SF6 0.06 MMS1-Kaufman et al., LOW SF6 17 11.52 -0.820.02 0.15 0.07 0.99 0.018713 2007 Western Williford et al. 2500 Mt. McRae Shale Australia MMS2 SIMS 24 14.83 -0.850.03 0.90 0.18 0.96 0 2016 Williford et al.. MMS2-UP SIMS 15 -1.030.06 0.72 0.15 0.95 0.000001 2016 8.14 MMS2-Williford et al. LOW SIMS 9 12.25 -0.880.03 1.39 0.24 0.99 2016 0 Kuruman Iron South Kaufman et al.. 2465 Africa KIF SF6 9 2.91 0.20 0.62 0.45 0.90 0.084866 2007 Formation -1.58South Johnson et al.. KF 0.132752 2415 Koegas Subgroup Africa SF6 24 2.01-1.260.19 0.16 0.14 0.68 2013

*P value for null hypothesis that the Δ^{36} S intercept is equal to zero. When P value shows as zero, means P value <0.000001.

* Uncertainties are expressed as 1 sigma (1SD).

- 1. Isotopologue-dependent SO₂ photoexcitation via collision-induced intersystem crossing at wavelengths of 240–340 nm. At low SO₂ partial pressures, appropriate to the Archean atmosphere, this results in a Δ^{36} S- Δ^{33} S fractionation line with a slope of +0.7 on a plot of Δ^{36} S against Δ^{33} S (Endo et al., 2016; Whitehill and Ono, 2012). At high SO₂ partial pressures, the experimental Δ^{36} S/ Δ^{33} S slopes can be up to +2.2 (Whitehill et al., 2013).
- 2. Isotopologue self-shielding in SO₂ photolysis at 190–220 nm wavelength, which produces negative Δ^{36} S- Δ^{33} S relationships, normally a mixture of mutual-shielding and ideal self-shielding with slopes that vary between -6.8 and -1.9, depending on the SO₂ partial pressures (e.g., Endo et al., 2016; Masterson et al., 2011; Ono, 2017; Ono et al., 2013; Whitehill and Ono, 2012). At low SO₂ partial pressures (pSO₂ < 10 Pa), appropriate for the Archean atmosphere, this photodissociation gives a slope of -2.4 (Endo et al., 2016).
- 3. Isotoplogue-dependent absorption cross-section at 190–220 nm wavelength, which produces a small S-MIF signal with $\Delta^{33}S$ varying by less than +9‰ but with no clear correlation between $\Delta^{33}S$ and $\Delta^{36}S$ (Endo et al., 2015).
- 4. Microbial sulfate reduction (MSR) produces the Phanerozoic MDF line with a Δ^{36} S/ Δ^{33} S slope of -6.9 and a Δ^{36} S range of -1.9‰ to -2.4‰ (Ono et al., 2006b).
- 5. The magnetic isotopic effect in thermochemical sulfate reduction (TSR), which fractionates $\Delta^{33}S$ without affecting $\Delta^{36}S$ (0 $<\Delta^{33}S\leq$ +13‰, $\Delta^{36}S\approx$ 0) (Oduro et al., 2011).

All of the above $\Delta^{36}S$ - $\Delta^{33}S$ correlation lines pass through the origin (Fig. 3) but none has a slope similar to the best-fit lines depicted in Figs. 1 and 2 for Archean samples. The challenge is to find a process, or combination of processes, that can produce the slopes and non-zero $\Delta^{36}S/\Delta^{33}S$ axis intercepts shown in Figs. 1 and 2 and documented in Table 1.

The carbon-rich nature of the host sediments to many of the Archean pyrites, NF-UP4A for example, are suggestive of biological activity. As a consequence, some biological fractionation of S isotopes is likely. However, MSR cannot explain the linear form of the best-fit lines for the NF-UP4A pyrites, nor their positive intercepts on the Δ^{36} S axis, for two reasons. First, all the analysed S from NF-UP4A have positive Δ^{33} S values (Fig. 1), which shows that the Δ^{33} S signal was derived dominantly from S⁰ and not SO₄²⁻. Second, MSR lowers Δ^{36} S but increases Δ^{33} S (Ono et al., 2006b), the effect on Δ^{36} S, however, is much greater than on Δ^{33} S (Fig. 3). As a consequence, MSR reduces, not increases, the intercept on the Δ^{36} S axis (Fig. 3). It can explain the negative intercepts on the Δ^{36} S axis for the barite-associated pyrites from the North Pole, Pilbara Craton, Western Australia (Ueno et al., 2008; Shen et al., 2009) and the Barberton Greenstone Belt, South Africa (Roerdink et al., 2013), where there is convincing evidence that MSR has been active but it cannot explain the positive Δ^{36} S intercepts seen in most Archean pyrite suites (Fig. 2).

Thermochemical sulfate reduction (TSR), like MSR requires sulfate reduction, which as already noted, is inconsistent with the positive Δ^{33} S values found in NF-UP4A and most other Archean sedimentary pyrite suites. Furthermore, TSR requires temperatures close to 300 °C (Oduro et al., 2011). The Upper Nauga Formation is a plus 200-m-thick C-rich marine sediment (Izon et al., 2017), so TSR could not be important during in situ pyrite precipitation at realistic ocean temperatures. It could, however, take place during sulfide formation in submarine black smokers by reduction of seawater sulfate entrained into the rising hydrothermal plume. Because the high rate of entrainment in turbulent plumes (Turner, 1973), leads to extreme thermal gradients, TSR would have to occur within approximately 10 cm of the chimney vent. Although possible under such circumstances TSR is unlikely to be important. Finally, TSR cannot explain the negative intercepts on the $\Delta^{36} S$ axis seen in some Archean pyrite suites, for example, the North Pole.

We now discuss an alternative explanation for the non-zero intercepts on the $\Delta^{36}S$ and $\Delta^{33}S$ axes observed in the NF-UP4A pyrites, and many other Archean examples.

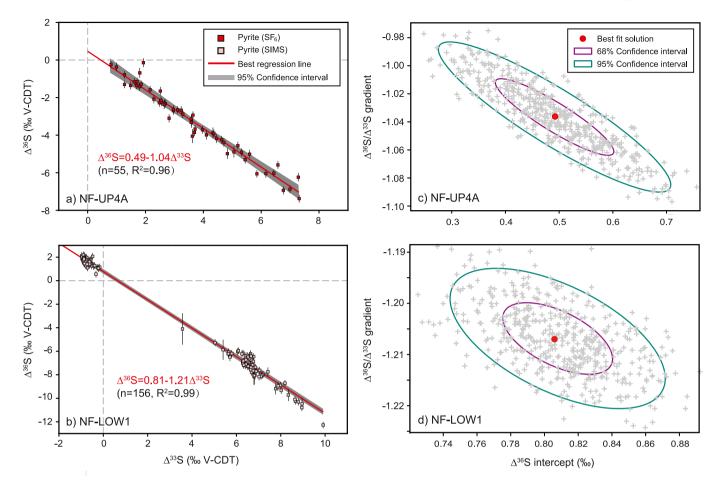


Fig. 1. a-b) Plots of Δ^{36} S vs. Δ^{33} S for two Archean sedimentary pyrite suites. Red squares represent the NF-UP4A pyrite suite, and pink squares the NF-LOW1 pyrite suite. One-sigma external uncertainties for Δ^{36} S, shown in panels a-b as vertical black lines. One-sigma external uncertainties for Δ^{33} S, which are much smaller than the data symbols, are not shown. The red solid lines are the best-fit linear regression lines for each data set, grey-shaded areas are the 95% confidence limits for the best-fit lines (Python code for least-squared linear regression fitting is given in *Supplementary materials*); c-d) Random draws of feasible solutions for the fitted Δ^{36} S/ Δ^{33} S gradient and Δ^{36} S intercept (‰) for NF-UP4A and NF-LOW1. Each cross represents a random draw from the model error distribution for the gradient-intercept pair. Red dots represent the mean of the distribution of gradient and intercept pairs, i.e., the red solid lines in panels a-b; blue and green ellipses are 68% and 95% confidence levels, respectively. The grey crosses within the green ellipses are the grey area in panels a-b.

4. Quadruple-sulfur isotope fractionation by simultaneous fractionation of SO_2

Atmospheric mixing, i.e., SO2 fractionation by two atmospheric processes known from experimental studies, has been suggested as a potential mechanism by several researchers (e.g., Whitehill and Ono, 2012; Endo et al., 2016; Mishima et al., 2017). Whitehill and Ono (2012) suggested "mixing the signatures of the 190-220 nm band and the 250-330 nm band in the right proportions could produce the Archean Δ^{36} S/ Δ^{33} S ratio of -0.9" but did not expand on this statement or attempt to show how the known experimental arrays can be combined to produce the observed Archean linear arrays. Endo et al. (2016) also argued for a mixed signature from the 190-220 nm and the 250-330 nm bands and suggested that a minimum of 3.3% S input from photoexcitation is required to produce the Δ^{36} S/ Δ^{33} S ratio of the ~2.5 Ga Mt. McRae Formation. Similarly, Mishima et al. (2017) suggested an atmospheric mixing model in which "the SO₂ self-shielding rate and photoexcitation rate grew after 2.7 Ga" to explain variations in $\Delta^{36}S/\Delta^{33}S$ ratios. These atmospheric mixing models, without exception, overlooked the pervasive existence of non-zero $\Delta^{36}\!S$ and $\Delta^{33}\!S$ intercepts in the Archean sedimentary records. Crucially, the observed Archean linear arrays cannot be the result of mixing of sulfur from two or more independent atmospheric reservoirs, as the resulting mixture would not form a straight line but scatter over a broad field, as will be discussed in greater

detail below.

Therefore, we propose an alternative explanation, which is consistent with the Archean sedimentary data, that the observed best-fit lines are the product of simultaneous isotopic fractionation of SO₂ by two or more of the atmospheric processes known from experimental studies. These SO₂ photochemical processes must occur simultaneously, in a single atmospheric reservoir, in order that the fraction contributed by the end-member process remains constant across the full range of Δ^{33} S values. The difference between mixing and simultaneous fractionation may appear to be semantic so we give simple illustration to explain the difference. If the fractionation of S isotopes in NF-UP4A were produced by S in one reservoir (Reservoir A), lying on the photoexcitation line, mixing with S from a second reservoir (Reservoir B) that was fractionated by the self-shielding photolysis process, the S isotopes in both reservoirs, prior to mixing, would be distributed along two straight lines as shown in Fig. 3. The S isotope distribution in both reservoirs would, of course, be heterogeneous. For illustrative purposes, we have assumed that the end-members of these linear reservoirs can be represented by A1 and A2, and B1 and B2 respectively. If mixed occurred within reservoir A the compositions produced could lie anywhere on the $A_1\mathchar`-A_2$ straight line, but the end members would converge and eventually meet if mixing progressed to the point where the mixture became homogeneous. The same logic applies to mixing with Reservoir B. Mixing between linear arrays A and B, however, would produce a scatter of data

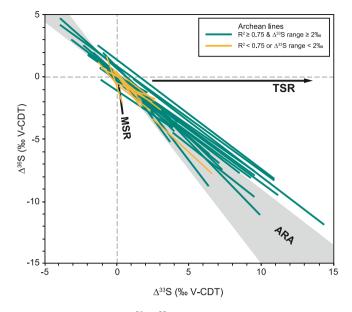


Fig. 2. The best-fit lines for Δ^{36} S- Δ^{33} S correlations of the Archean sedimentary pyrite suites. The range of Δ^{36} S/ Δ^{33} S gradients for the Archean Reference Array (ARA) (Ono, 2017), the Phanerozoic microbial sulfate reduction (MSR) massdependent fractionation line (Ono et al., 2006b), and thermochemical sulfate reduction line (TSR) (Oduro et al., 2011), are also plotted. Over half of the bestfit lines for Archean suites, with high-reliability (green lines, $R^2 \ge 0.75$ and Δ^{33} S ranges $\geq 2\%$), have positive Δ^{36} S intercepts. Archean suites with low reliability (yellow lines, $R^2 < 0.75$ or $\Delta^{33}S$ ranges < 2‰) are also shown for comparison. The data for linear regression lines are from Busigny et al. (2017); Thomassot et al. (2015); Montinaro et al. (2015); Diekrupa et al. (2018); Gallagher et al. (2017); Mishima et al. (2017); Shen et al. (2009); Ueno et al. (2008); Roerdink et al. (2013); Thomazo et al. (2009); Farquhar et al. (2007); Kurzweil et al. (2013); Zhelezinskaia et al. (2014); Zerkle et al. (2012); Williford et al. (2016); Thomazo et al. (2013); Farquhar et al. (2013); Ono et al. (2009); Kaufman et al. (2007) and Johnson et al. (2013). Four Archean sedimentary suites showing no statistically reliable linear trend ($R^2 < 0.50$), including those from Ono et al. (2006a), Wacey et al. (2015), Williford et al. (2016) and Saitoh et al. (2020), are not shown. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

lying between the four end member reservoirs, that is a quadrilateral, as illustrated in Fig. 3. Mixing between endmember A_1 and B_1 can produce points that lie anywhere on the line between A_1 and B_1 , mixing between endmembers A_1 and B_2 , will produce points that lie on the line between A_1 and B_2 , etc. A linear array can only be preserved if mixing occurs within a single linear array. The two fractionation processes must occur simultaneously in a single reservoir *–simultaneous fractionation–* as is expected when atmospheric SO₂ is exposed to natural light with a continuum of wavelengths.

The model we have used to constrain how the experimental arrays can combine to produce the observed Archean linear arrays is illustrated in Fig. 4. We accept the arguments of Endo et al. (2016) that the 190–220 nm wavelength experiments, carried out prior to theirs, were run at inappropriately high pSO₂ (>1000 Pa), equivalent to >10,000 ppm SO₂ (e.g., Lyons, 2007, 2008). This compares with 10 ppb SO₂ in the modern atmosphere and a few ppm in volcanic plumes. Furthermore, the earlier experiments were carried out under conditions that were appreciably more oxidizing than those appropriate for the Archean atmosphere. We have therefore used the slope of -2.4 for $\Delta^{36}S-\Delta^{33}S$ fractionation by SO₂ photodissociation with the self-shielding effect in the 190–220 nm wavelength range and +0.7 for photoexcitation in the 240–340 nm range as recommended by Endo et al. (2016).

The equations that characterise the $\Delta^{36}S \cdot \Delta^{33}S$ correlation lines have the form Y = kX + C, where Y is $\Delta^{36}S$, X is $\Delta^{33}S$, k is the gradient and C is

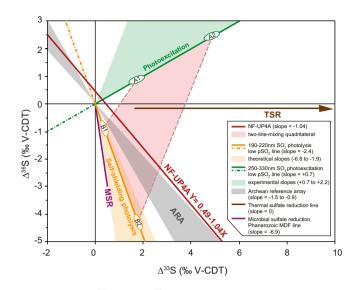


Fig. 3. A plot of Δ^{36} S against Δ^{33} S showing the trend lines for SO₂ photoexcitation (Endo et al., 2016), isotopologue self-shielding photolysis (Endo et al., 2016), microbial sulfate reduction (MSR) (Ono et al., 2006b), and thermochemical sulfate reduction (TSR) (Oduro et al., 2011). The dashed green and yellow lines are the extension of the photochemical experiment lines to negative Δ^{33} S. The green and orange shaded areas show the slope ranges of the photochemical experiment lines under different SO₂ partial pressure (Endo et al., 2016; Masterson et al., 2011; Ono et al., 2013; Whitehill et al., 2013; Whitehill and Ono, 2012). The grey shaded area shows the Archean Reference Array (ARA) (Ono, 2017). Note that the best-fit line for the NF-UP4A data has a positive intercept on the Δ^{36} S and that none of the experimental fractionation lines have gradients that are close to that of NF-UP4A. The ellipses marked with A1, A2, and B1, B2, represent end member reservoirs of linear arrays, produced by the two different photochemical reactions. The pale red shaded quadrilateral represents the line-to-line mixing, which cannot produce a linear array. The NF-UP4A trend could be produced by a combination of SO₂ photoexcitation and self-shielding photolysis if the two reactions occurred simultaneously in a single reservoir. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

the intercept on the y ($\Delta^{36}S$) axis. Y = kX + C can be reproduced by combining $Y_1 = k_1X_1$ with $Y_2 = k_2X_2$, if $k_1 > k > k_2$, where (x_1, y_1) is a point on Line 1, and (x_2, y_2) is a point on Line 2. $(x_1 + L_1)$ is the X coordinate of a second point on Line 1 and $(x_2 + L_2)$ is the X coordinate of a second point on Line 2. Since photoexcitation of SO₂ is the only process that yields an array with a gradient > k on a plot of $\Delta^{36}S \vee \Delta^{33}S$, k_1 must be +0.7. The process that yields a gradient < k is photodissociation with self-shielding, so that $k_2 = -2.4$. X and Y for the best-fit lines are given by:

$$X = fX_1 + (1-f) X_2$$
 (1)

$$Y = fk_1X_1 + (1-f)k_2X_2$$
(2)

The fraction (f) is defined as the fraction of S from the photoexcitation fractionation line that contributes to the combined line:

$$f = L_2(k - k_2) / \{L_1(k_1 - k) + L_2(k - k_2)\}$$
(3)

The intercept C is given by.

$$C = (k_1 - k) X_1 f - (k - k_2) (1 - f) X_2$$
(4)

If $L_2 = L_1$, $f = (k - k_2)/(k_1 - k_2)$, the mixing solution is unique.

Because x_1 and x_2 are linearly correlated with y_1 and y_2 , respectively, it can be shown from Eqs. (3) and (4) that when $x_1/L_1 = x_2/L_2$ the combination line (Y = kX + C) passes through the origin, whereas if $x_1/L_1 \neq x_2/L_2$, it does not. If $x_1/L_1 > x_2/L_2$ the line has a positive intercept on the y-axis but a negative intercept if $x_1/L_1 < x_2/L_2$. If $x_1/L_1 > x_2/L_2$ the positive intercept increases as $(x_1/L_1)/(x_2/L_2)$ increases and if x_1/L_1

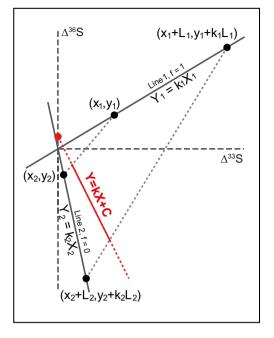


Fig. 4. Diagram used to illustrate the model described in the text to produce the observed $\Delta^{36}S$ vs. $\Delta^{33}S$ fractionation lines by combining SO₂ photoexcitation with SO₂ self-shielding photolysis. Line 1 represents SO₂ photoexcitation and Line 2 SO₂ self-shielding photolysis. The line to be modelled is Y=kX+C.

 $< x_2/L_2$ the negative intercept increases with decreasing (x_1/L_1)/(x_2/L_2).

There is a continuum of solutions if $L_2 \neq L_1$. This continuum has been modelled using python (see Supplementary materials), assuming a single value of f for combining the possible end members. As noted above k2 (photodissociation with self-shielding) has been set at -2.4 and the range of Δ^{33} S (in per mil) used is 0 to 22, which is consistent with the experimentally observed range of Δ^{33} S reported by Endo et al. (2016). The $\Delta^{36}S/\Delta^{33}S$ for photoexcitation (k₁) is assumed to be 0.7. Values for f have been calculated for three ranges of Δ^{33} S (in per mil), based on experimentally observed ranges, to evaluate their influence on f: Model 1, 0-142 (Ono, 2017); Model 2, 13-142 (Endo et al., 2016; Whitehill and Ono, 2012; Whitehill et al., 2013); and Model 3, 118-142 (Endo et al., 2016). In all three models, x_1 was varied from the start of the range to the end in increments of 0.2, and the length ranges of Δ^{33} S for L₁ and L₂. were varied from the maximum value as follows: Model 1, L1 in increment of -4 and L_2 in increment of -0.4; Model 2, L_1 in increment of -2and L_2 in increment of -0.2; and in Model 3, both L_1 and L_2 were varied in increment of -0.2. As a consequence, all possible combinations of x_1 , L1, and L2 were canvassed within the stated limits. The solution frequency of f for NF-UP4A is shown in Fig. 5. Note that the mode and medium values for f are about 0.09 in Models 1 and 2 but reduce to 0.06 for Model 3. Halving the X₂ range to 0–11, while maintaining the range of X_1 , halves f whereas halving the range of X_1 while maintaining X_2 doubles the value of f. Extending the range of Δ^{33} S to negative values (-4) in Model 1, for example, has little influence on the calculated values for f (see Model 4, Fig. 5).

The mean, mode, median, minimum, and maximum solutions for f are listed in Table 2. Values for the median and mean, take NF-UP4A for example, are similar: with the mean > median for Models 1, 2, and 4 but the mean slightly less than the median for Model 3 (Table 2, Fig. 5). It is worth noting that the solutions for the modelled combined fractionation line could have variable Δ^{33} S magnitudes and ranges, and as expected, some exceed those observed in the Archean sedimentary suites (Table A.1) due to high Δ^{33} S magnitudes and wide Δ^{33} S ranges from the photochemical experiments. During transfer and deposition of atmospheric sulfur, MIF signals can be diluted by mixing of atmospheric

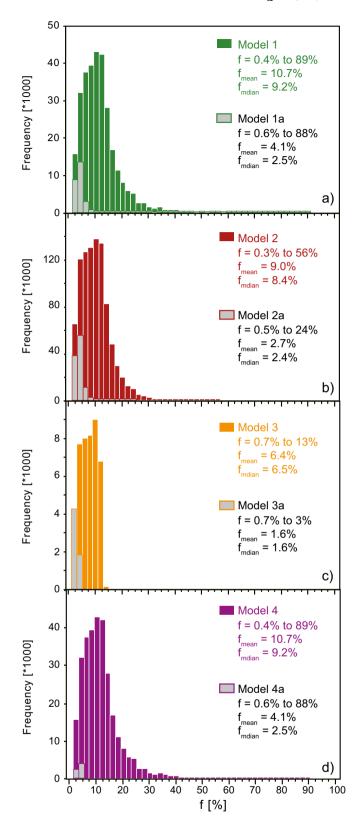


Fig. 5. Solution frequency of fraction (f) for NF-UP4A for Models 1–4 (coloured filled bins) and Models 1a-4a (grey filled bins). For Models 1a-4a, the observed Δ^{33} S magnitudes and ranges of NF-UP4A were taken into account. The fraction (f) is defined as the fraction of S from the SO₂ photoexcitation fractionation line that contributes to the simultaneous fractionation line. See text for detailed description.

Table 2 Statistics of the fraction f for the compiled Archean sedimentary pyrites.

Sample ID	$\Delta 36S/\Delta 33S$	Δ^{36} S intercept	Model 1 (0-22 vs 0-142)					Model 2 (0-22 vs 13-142)					Model 3 (0-22 vs 118-142)				
			f _{min}	\mathbf{f}_{\max}	f _{mean}	f _{median}	f _{mode}	f _{min}	\mathbf{f}_{\max}	f _{mean}	f _{median}	\mathbf{f}_{mode}	f _{min}	\mathbf{f}_{\max}	f _{mean}	f _{median}	\mathbf{f}_{mode}
NGB	-0.853	-0.623	0.003	0.898	0.117	0.103	0.117	0.002	0.582	0.102	0.097	0.117	0.008	0.15	0.077	0.078	0.125
DF1	-1.367	-0.592	0.001	0.821	0.065	0.054	0.062	0.001	0.411	0.055	0.050	0.062	0.004	0.080	0.040	0.040	0.066
DF2	-0.808	-0.205	0.003	0.915	0.125	0.110	0.123	0.002	0.596	0.109	0.103	0.174	0.009	0.157	0.082	0.083	0.131
DF3	no trend																
MNF	-0.467	-0.654	0.005	0.933	0.174	0.161	0.181	0.003	0.699	0.155	0.151	0.181	0.014	0.226	0.12	0.124	0.191
MF_BARITE	-3.558	-0.704															
MF_CHERT	-0.874	0.080	0.003	0.909	0.119	0.105	0.115	0.002	0.577	0.103	0.097	0.163	0.008	0.151	0.077	0.078	0.122
MF SHALE	-0.635	-0.425	0.004	0.924	0.148	0.134	0.15	0.002	0.649	0.13	0.125	0.209	0.011	0.189	0.099	0.102	0.159
MDG	no trend																
BG	-1.474	-0.030	0.001	0.822	0.059	0.048	0.054	0.001	0.375	0.049	0.045	0.054	0.004	0.072	0.036	0.036	0.058
VF	-1.171	-0.130	0.002	0.872	0.085	0.072	0.116	0.001	0.479	0.072	0.067	0.116	0.005	0.107	0.053	0.054	0.086
MG	no trend																
TGB	-0.737	0.042	0.004	0.92	0.136	0.122	0.134	0.002	0.618	0.119	0.113	0.188	0.01	0.174	0.089	0.091	0.142
TGB BIF	-1.377	0.351	0.003	0.831	0.071	0.06	0.062	0.002	0.441	0.059	0.054	0.062	0.004	0.082	0.041	0.042	0.066
TGB CHERT	-0.632	0.022	0.004	0.93	0.151	0.137	0.15	0.002	0.65	0.133	0.127	0.21	0.011	0.194	0.101	0.103	0.159
TGB_VOLC	-0.567	0.029	0.004	0.935	0.161	0.147	0.162	0.002	0.669	0.142	0.137	0.224	0.012	0.208	0.109	0.112	0.171
TGB_SHALE	-0.899	0.342	0.004	0.904	0.12	0.106	0.111	0.003	0.6	0.103	0.096	0.158	0.008	0.146	0.075	0.076	0.118
TF	-1.612	0.232	0.002	0.784	0.051	0.042	0.044	0.001	0.354	0.042	0.038	0.044	0.003	0.059	0.029	0.029	0.047
KCRC	-0.928	0.116	0.002	0.902	0.113	0.099	0.107	0.001	0.558	0.097	0.091	0.153	0.008	0.141	0.029	0.073	0.114
IF	-0.770	-1.074	0.003	0.896	0.115	0.112	0.129	0.002	0.608	0.110	0.105	0.133	0.009	0.156	0.072	0.085	0.137
BF	-0.928	0.200	0.003	0.906	0.120	0.112	0.129	0.002	0.559	0.098	0.103	0.152	0.009	0.130	0.072	0.033	0.137
JF	-0.887	0.888	0.004	0.905	0.113	0.11	0.160	0.002	0.603	0.098	0.103	0.155	0.008	0.141	0.072	0.073	0.114
LF	-1.016	0.281	0.007	0.903	0.129	0.091	0.100	0.003	0.562	0.089	0.103	0.138	0.008	0.147	0.065	0.078	0.119
HF	-0.951	-0.309	0.004	0.897	0.105	0.091	0.105	0.002	0.552	0.089	0.083	0.138	0.007	0.127	0.069	0.000	0.103
CD		-0.309	0.003	0.890	0.107	0.093	0.105	0.001	0.552	0.093	0.087	0.149	0.007	0.134	0.009	0.070	0.112
CF	no trend	0.264	0.002	0.769	0.048	0.039	0.040	0.001	0.334	0.039	0.035	0.040	0.003	0.054	0.027	0.027	0.043
	-1.658		0.002			0.039				0.039	0.035		0.005				0.043
NF_LOW1	-1.207	0.806		0.867	0.092		0.111	0.004	0.5			0.111		0.105	0.052	0.053	
NF_LOW2	-0.977	-0.008	0.002	0.901	0.105	0.091	0.101	0.001	0.542	0.091	0.085	0.145	0.007	0.133	0.067	0.068	0.108
NF_UP1	-0.974	-0.298	0.002	0.895	0.105	0.091	0.146	0.001	0.544	0.09	0.085	0.146	0.007	0.131	0.067	0.068	0.108
NF_UP2	-1.074	-0.654	0.002	0.871	0.093	0.080	0.091	0.001	0.513	0.080	0.074	0.131	0.006	0.117	0.059	0.060	0.097
NF_UP3	-0.897	-0.089	0.003	0.909	0.114	0.100	0.111	0.001	0.568	0.099	0.093	0.158	0.008	0.146	0.074	0.075	0.118
NF_UP4H	-1.579	1.144	0.005	0.783	0.062	0.051	0.046	0.004	0.366	0.052	0.047	0.046	0.004	0.064	0.032	0.033	0.049
NF_UP4A	-1.036	0.492	0.004	0.891	0.107	0.092	0.095	0.003	0.557	0.090	0.084	0.136	0.007	0.125	0.064	0.065	0.101
GF1	-0.855	0.048	0.003	0.908	0.120	0.106	0.116	0.002	0.580	0.104	0.098	0.165	0.008	0.152	0.078	0.079	0.124
GF2	-1.001	-0.414	0.002	0.888	0.101	0.087	0.099	0.001	0.535	0.087	0.081	0.141	0.007	0.127	0.065	0.066	0.105
GF2-STEEP	-1.189	0.319	0.003	0.874	0.088	0.075	0.114	0.002	0.506	0.074	0.068	0.114	0.005	0.104	0.053	0.054	0.084
GF2-SHALLOW	-0.914	-0.994	0.003	0.88	0.108	0.095	0.109	0.001	0.563	0.094	0.089	0.109	0.008	0.133	0.071	0.071	0.116
KNF	-1.007	0.251	0.003	0.896	0.106	0.092	0.140	0.002	0.565	0.090	0.084	0.140	0.007	0.129	0.065	0.067	0.104
MMS1	-0.849	0.173	0.004	0.912	0.124	0.109	0.118	0.002	0.584	0.106	0.1	0.167	0.008	0.154	0.079	0.081	0.125
MMS1_UP	-1.021	0.296	0.004	0.896	0.105	0.092	0.097	0.002	0.562	0.089	0.083	0.138	0.007	0.127	0.065	0.066	0.103
MMS1_LOW	-0.815	0.154	0.004	0.916	0.128	0.113	0.122	0.002	0.594	0.11	0.104	0.173	0.009	0.16	0.082	0.084	0.13
MMS2	-0.845	0.899	0.007	0.909	0.135	0.118	0.167	0.006	0.615	0.115	0.108	0.167	0.008	0.154	0.080	0.082	0.125
MMS2_UP	-1.034	0.716	0.005	0.890	0.110	0.096	0.137	0.004	0.559	0.093	0.087	0.137	0.007	0.126	0.065	0.066	0.102
MMS2_LOW	-0.877	1.386	0.009	0.912	0.136	0.120	0.114	0.008	0.606	0.118	0.110	0.161	0.008	0.153	0.079	0.080	0.121
KIF	-1.582	0.622	0.003	0.792	0.058	0.047	0.046	0.002	0.365	0.047	0.043	0.046	0.003	0.063	0.031	0.032	0.049
KF	-1.260	0.157	0.002	0.858	0.079	0.067	0.104	0.001	0.449	0.066	0.061	0.104	0.005	0.096	0.048	0.048	0.077

sulfur carrying opposite larger MIF anomalies, by S derived from MDF processes. The influence of these processes, though poorly known and difficult to quantify, indicates an unaffected initial atmospheric fractionation line with larger Δ^{33} S magnitudes and wider Δ^{33} S ranges. To obtain a minimum estimation of f values for the Archean sedimentary suites, the sulfur contents of the Archean sedimentary pyrites were assumed to consist of purely atmospheric sulfur. A constraint that the f solutions must satisfy the observed Δ^{33} S distributions was then added. New f solution sets of models 1-4 for NF-UP4A and the Mt. McRae were tested. In the case of NF-UP4A (Models 1a-4a in Fig. 5), the f solutions of models with or without constraint show similar distributions, but total numbers and magnitudes of mean and median solutions for f obtained at the given constraint, decrease significantly by \sim 7%, from around 11% and 9% to 4% and 2% respectively. Modelling of the Mt. McRae, which has similar $\Delta^{36}S/\Delta^{33}S$ slope to that of NF-UP4A, show a decrease in f values with mean and median values down to 2-3% and 1-2%, consistent with the estimation of Endo et al. (2016). Fig. 6 shows a plot of the mean and medium solutions for f against the Δ^{36} S/ Δ^{33} S gradient for the data taken from the literature. There is a quadratic polynomial relationship between $\Delta^{36}S/\Delta^{33}S$ and f for the mean and medium for the highest probability solutions (Fig. 6).

As discussed, the process that has been modelled is not an atmospheric mixing (Mishima et al., 2017; Whitehill and Ono, 2012) as might occur, for example, if the S on Line 1 was fractionated at one level in the atmosphere and that on Line 2 at another, and these two reservoirs subsequently mixed. The resulting mixture would not form a straight line but scatter over a broad field between Lines 1 and 2. The model

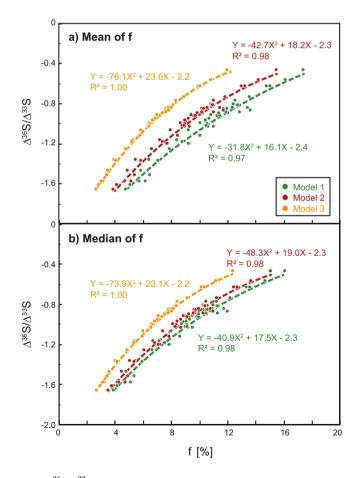


Fig. 6. $\Delta^{36}S/\Delta^{33}S$ plotted against f for the mean (a) and median (b) solutions within three different L ranges for all datasets. The fraction (f) is defined as the fraction of S from the SO₂ photoexcitation fractionation line that contributes to the combined line. There is a quadratic polynomial relationship between $\Delta^{36}S/\Delta^{33}S$ and f for the mean and medium values for all three classes of solution.

illustrated in Fig. 4 is only viable if the S on Lines 1 and 2 is fractionated simultaneously by two processes in a single reservoir. This is possible because SO₂ has two principal absorption bands within the ultraviolet bandwidth, 240–340 nm and 190–220 nm. As noted earlier, photoexcitation of SO₂ occurs in the former bandwidth, and photodissociation with self-shielding in the later. Since natural sunlight includes both bandwidths, photodissociation with self-shielding and photoexcitation via collision-induced intersystem crossing can occur simultaneously, within the same reservoir, as required by our hypothesis. Our model predicts the relative contributions of photoexcitation and photodissociation required to produce the observed Δ^{36} S- Δ^{33} S linear arrays and Δ^{36} S intercepts.

5. Potential factors controlling the Archean $\bigtriangleup^{36}\text{S}/\bigtriangleup^{33}\text{S}$ distribution

Changes in $\Delta^{36}S/\Delta^{33}S$ with time have been attributed to variations in atmospheric chemistry, including: (i) changes in CH_4 (organic haze: Domagal-Goldman et al., 2008; Izon et al., 2015; Zerkle et al., 2012), (ii) fluctuation in oxygen (Ohmoto et al., 2006; Ono et al., 2006a), (iii) changes in SO₂/H₂S (Halevy et al., 2010), and (iv) increased selfshielding effect due to enhanced SO₂ partial pressure (Ono et al., 2013). Our study shows that variations in f control changes in Δ^{36} S/ Δ^{33} S but changes in f could, in turn, be controlled by changes in atmospheric chemistry. It is also worth noting that f values might change with altitude as self-shielding photolysis is expected to be dependent on the altitude (Ono et al., 2003). The problem is too poorly constrained to provide a definitive answer but enhanced SO₂ partial pressure is the most likely candidate because it controls photodissociation with selfshielding and therefore influences f. The plot of $\Delta^{36}S/\Delta^{33}S$ against time shows no evidence of a systematic relationship between these variables (Fig. 7). The observed variations in Δ^{36} S/ Δ^{33} S may be due to global changes in the partial pressure of SO₂ in the atmosphere in response to the waxing and waning of subaerial volcanic activity.

6. Conclusions

The following conclusions can be drawn from this study:

- The best-fit lines for plots of Δ³⁶S vs. Δ³³S for many Archean sedimentary pyrite suites do not pass through the origin, requiring the fractionation of Δ³⁶S and Δ³³S to involve at least two processes.
- 2. Two processes that can produce the observed Archan $\Delta^{36}S/\Delta^{33}S$ gradients are SO₂ photoexcitation ($\Delta^{36}S/\Delta^{33}S = +0.7$), which occurs in the 240–340 nm bandwidth, and SO₂ photodissociation ($\Delta^{36}S/\Delta^{33}S = -2.4$), which takes place in the 190–220 nm bandwidth.
- 3. The observed linear relationship between $\Delta^{36}S$ and $\Delta^{33}S$ requires the contribution from the two end-member processes to be in the same proportion over a wide range of $\Delta^{33}S$ values, which precludes mixing between different atmospheric reservoirs of S.
- 4. However, it can be explained if both photoexcitation via collisioninduced intersystem crossing, and photodissociation with selfshielding, occur simultaneously in a single reservoir as is expected when atmospheric SO₂ is exposed to natural light. We suggest this process should be called *simultaneous fractionation*.
- 5. Numerical modelling shows that variations in $\Delta^{36}S/\Delta^{36}S$ gradients are controlled by changes in f, the fraction of photoexcitation S that contributes to the simultaneous fractionation line. Increasing f increases $\Delta^{36}S/\Delta^{36}S$ gradients.
- 6. Variations in $\Delta^{36}S/\Delta^{36}S$ gradients are most simply explained by changes in the partial pressure of SO₂ in the atmosphere, which control the relative contributions from photodissociation and photoexcitation to simultaneous fractionation and therefore f.

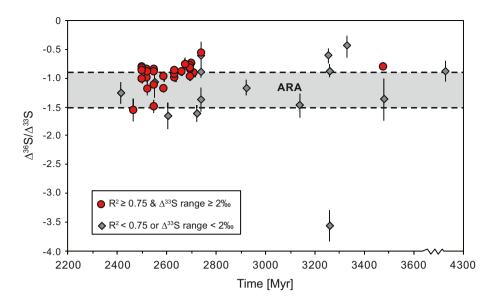


Fig. 7. $\Delta^{36}S/\Delta^{33}S$ plotted against time for the Archean sedimentary pyrites. The grey zone defined by two dashed black lines is the $\Delta^{36}S/\Delta^{33}S$ range of the Archean Reference Array suggested in Ono (2017). Red dots show the gradients for best-fit lines for Archean suites with high reliability ($R^2 \ge 0.75$ and $\Delta^{33}S$ ranges $\ge 2\%$), and grey diamonds for those with low reliability ($R^2 < 0.75$ or $\Delta^{33}S$ ranges <2‰). The error bars are 1-sigma standard deviations for $\Delta^{36}S/\Delta^{33}S$ gradients.

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Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.chemgeo.2021.120640.

References

- Busigny, V., Marin-Carbonne, J., Muller, E., Cartigny, P., Rollion-Bard, C., Assayag, N., Philippot, P., 2017. Iron and sulfur isotope constraints on redox conditions associated with the 3.2 Ga barite deposits of the Mapepe Formation (Barberton Greenstone Belt, South Africa). Geochim. Cosmochim. Acta 210, 247–266.
- Diekrupa, D., Hanningtona, M.D., Strauss, H., Ginley, S.J., 2018. Decoupling of Neoarchean sulfur sources recorded in Algoma-type banded iron formation. Earth Planet. Sci. Lett. 489, 1–7.
- Domagal-Goldman, S.D., Kasting, J.F., Johnston, D.T., Farquhar, J., 2008. Organic haze, glaciations and multiple sulfur isotopes in the Mid-Archean Era. Earth Planet. Sci. Lett. 269, 29–40.
- Endo, Y., Danielache, S.O., Ueno, Y., Hattori, S., Johnson, M.S., Yoshida, N., Kjaergaard, H.G., 2015. Photoabsorption cross-section measurements of ³²S, ³³S, ³⁴S and ³⁶S sulfur dioxide from 190 to 220 nm. J. Geophys. Res. Atmos. 120, 2546–2557.
- Endo, Y., Ueno, Y., Aoyama, S., Danielache, S.O., 2016. Sulfur isotope fractionation by broadband UV radiation to optically thin SO₂ under reducing atmosphere. Earth Planet. Sci. Lett. 453, 9–22.
- Farquhar, J., Bao, H., Thiemens, M., 2000. Atmospheric Influence of Earth's Earliest Sulfur Cycle. Science 289, 756–758.
- Farquhar, J., Savarino, J., Airieau, S., Thiemens, M.H., 2001. Observation of wavelengthsensitive mass-independent sulfur isotope effects during SO₂ photolysis: Implications for the early atmosphere. J. Geophys. Res. Planets 106, 32829–32839.

- Farquhar, J., Johnston, D.T., Wing, B.A., Habicht, K.S., Canfield, D.E., Airieau, S., Thiemens, M.H., 2003. Multiple Sulphur isotopic interpretations of biosynthetic pathways: implications for biological signatures in the Sulphur isotope record. Geobiology 1, 27–36.
- Farquhar, J., Peters, M., Johnston, D.T., Strauss, H., Masterson, A., Wiechert, U., Kaufman, A.J., 2007. Isotopic evidence for Mesoarchaean anoxia and changing atmospheric Sulphur chemistry. Nature 449, 706–709.
- Farquhar, J., Cliff, J., Zerkle, A.L., Kamyshny, A., Poulton, S.W., Claire, M., Adams, D., Harms, B., 2013. Pathways for Neoarchean pyrite formation constrained by massindependent sulfur isotopes. Proc. Natl. Acad. Sci. U. S. A. 110, 17638–17643.
- Galić, A., Mason, P.R.D., Mogollón, J.M., Wolthers, M., Vroon, P.Z., Whitehouse, M.J., 2017. Pyrite in a sulfate-poor Paleoarchean basin was derived predominantly from elemental sulfur: evidence from 3.2 Ga sediments in the Barberton Greenstone Belt, Kaapvaal Craton. Chem. Geol. 449, 135–146.
- Gallagher, M., Whitehouse, M.J., Kamber, B.S., 2017. The Neoarchaean surficial Sulphur cycle: an alternative hypothesis based on analogies with 20th-century atmospheric lead. Geobiology 15, 385–400.
- Halevy, I., Johnston, D.T., Schrag, D.P., 2010. Explaining the Structure of the Archean Mass-Independent Sulfur Isotope Record. Science 329, 204.
- Izon, G., Zerkle, A.L., Zhelezinskaia, I., Farquhar, J., Newton, R.J., Poulton, S.W., Eigenbrode, J.L., Claire, M.W., 2015. Multiple oscillations in Neoarchaean atmospheric chemistry. Earth Planet. Sci. Lett. 431, 264–273.
- Izon, G., Zerkle, A.L., Williford, K.H., Farquhar, J., Poulton, S.W., Claire, M.W., 2017. Biological regulation of atmospheric chemistry en route to planetary oxygenation. Proc. Natl. Acad. Sci. U. S. A. 114, E2571–E2579.
- Johnson, J.E., Webb, S.M., Thomas, K., Ono, S., Kirschvink, J.L., Fischer, W.W., 2013. Manganese-oxidizing photosynthesis before the rise of cyanobacteria. Proc. Natl. Acad. Sci. U. S. A. 110, 11238–11243.
- Johnston, D.T., Farquhar, J., Canfield, D.E., 2007. Sulfur isotope insights into microbial sulfate reduction: when microbes meet models. Geochim. Cosmochim. Acta 71, 3929–3947.
- Kaufman, A.J., Johnston, D.T., Farquhar, J., Masterson, A.L., Lyons, T.W., Bates, S., Anbar, A.D., Arnold, G.L., Garvin, J., Buick, R., 2007. Late Archean biospheric oxygenation and atmospheric evolution. Science 317, 1900–1903.
- Kurzweil, F., Claire, M., Thomazo, C., Peters, M., Hannington, M., Strauss, H., 2013. Atmospheric sulfur rearrangement 2.7 billion years ago: evidence for oxygenic photosynthesis. Earth Planet. Sci. Lett. 366, 17–26.
- Labidi, J., Cartigny, P., Hamelin, C., Moreira, M., Dosso, L., 2014. Sulfur isotope budget (³²S, ³³S, ³⁴S and ³⁶S) in Pacific-Antarctic ridge basalts: a record of mantle source heterogeneity and hydrothermal sulfide assimilation. Geochim. Cosmochim. Acta 133, 47–67.
- Lyons, J.R., 2007. Mass-independent fractionation of sulfur isotopes by isotope-selective photodissociation of SO₂. Geophys. Res. Lett. 34, 1–5.
- Lyons, J.R., 2008. Photolysis of long-lived predissociative molecules as a source of massindependent isotope fractionation: the example of SO₂. Adv. Quantum Chem. 55, 57–74.
- Masterson, A.L., Farquhar, J., Wing, B.A., 2011. Sulfur mass-independent fractionation patterns in the broadband UV photolysis of sulfur dioxide: pressure and third body effects. Earth Planet. Sci. Lett. 306, 253–260.
- Mishima, K., Yamazaki, R., Satish-Kumar, M., Ueno, Y., Hokada, T., Toyoshima, T., 2017. Multiple sulfur isotope geochemistry of Dharwar Supergroup, Southern India: late Archean record of changing atmospheric chemistry. Earth Planet. Sci. Lett. 464, 69–83.

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Montinaro A, Strauss, H., Mason, P. R. D., Roerdink, D., Münker, C., Schwarz-Schampera, U., Arndt, N. T., Farquhar, J., Beukes, N. J., Gutzmer, J. and Peters, M. (2015) Paleoarchean sulfur cycling: Multiple sulfur isotope constraints from the Barberton Greenstone Belt, South Africa. Precambrian Res. 267, 311–322.

- Muller, E., Philippot, P., Rollion-Bard, C., Cartigny, P., Assayag, N., Marin-Carbonne, J., Mohan, M.R., Sarma, S., 2017. Primary sulfur isotope signatures preserved in highgrade Archean barite deposits of the Sargur Group, Dharwar Craton. India. Precambrian Res. 295, 38–47.
- Oduro, H., Harms, B., Sintim, H.O., Kaufman, A.J., Cody, G., Farquhar, J., 2011. Evidence of magnetic isotope effects during thermochemical sulfate reduction. Proc. Natl. Acad. Sci. U. S. A. 108, 17635–17638.
- Ohmoto, H., Watanabe, Y., Ikemi, H., Poulson, S.R., Taylor, B.E., 2006. Sulphur isotope evidence for an oxic Archaean atmosphere. Nature 442, 908–911.
- Ono, S., 2017. Photochemistry of Sulfur Dioxide and the Origin of Mass-Independent Isotope Fractionation in Earth's Atmosphere. In: Jeanloz, R., Freeman, K.H. (Eds.), Annual Review of Earth and Planetary Sciences, vol. 45. Annual Reviews, Palo Alto, pp. 301–329.
- Ono, S., Eigenbrode, J.L., Pavlov, A.A., Kharecha, P., Rumble III, D., Kasting, J.F., Freeman, K.H., 2003. New insights into Archean sulfur cycle from mass-independent sulfur isotope records from the Hamersley Basin, Australia. Earth Planet. Sci. Lett. 213, 15–30.
- Ono, S., Beukes, N.J., Rumble, D., Fogel, M.L., 2006a. Early evolution of atmospheric oxygen from multiple-sulfur and carbon isotope records of the 2.9 Ga Mozaan Group of the Pongola Supergroup, Southern Africa. S. Afr. J. Geol. 109, 97–108.
- Ono, S., Wing, B., Johnston, D., Farquhar, J., Rumble, D., 2006b. Mass-dependent fractionation of quadruple stable sulfur isotope system as a new tracer of sulfur biogeochemical cycles. Geochim. Cosmochim. Acta 70, 2238–2252.
- Ono, S., Beukes, N.J., Rumble, D., 2009. Origin of two distinct multiple-sulfur isotope compositions of pyrite in the 2.5 Ga Klein Naute Formation, Griqualand West Basin, South Africa. Precambrian Res. 169, 48–57.
- Ono, S., Whitehill, A.R., Lyons, J.R., 2013. Contribution of isotopologue self-shielding to sulfur mass-independent fractionation during sulfur dioxide photolysis. J. Geophys. Res. Atmos. 118, 2444–2454.
- Roerdink, D.L., Mason, P.R.D., Whitehouse, M.J., Reimer, T., 2013. High-resolution quadruple sulfur isotope analyses of 3.2 Ga pyrite from the Barberton Greenstone Belt in South Africa reveal distinct environmental controls on sulfide isotopic arrays. Geochim. Cosmochim. Acta 117, 203–215.
- Roerdink, D.L., Mason, P.R.D., Whitehouse, M.J., Brouwer, F.M., 2016. Reworking of atmospheric sulfur in a Paleoarchean hydrothermal system at Londozi, Barberton Greenstone Belt, Swaziland. Precambrian Res. 280, 195–204.
- Saitoh, M., Nabhan, S., Thomazo, C., Olivier, N., Moyen, J.F., Ueno, Y., Marin-Carbonne, J., 2020. Multiple Sulfur Isotope Records of the 3.22 Ga Moodies Group, Barberton Greenstone Belt. Geosciences 10, 145.

- Shen, Y., Farquhar, J., Masterson, A., Kaufman, A.J., Buick, R., 2009. Evaluating the role of microbial sulfate reduction in the early Archean using quadruple isotope systematics. Earth Planet. Sci. Lett. 279, 383–391.
- Thomassot, E., O'Neil, J., Francis, D., Cartigny, P., Wing, B.A., 2015. Atmospheric record in the Hadean Eon from multiple sulfur isotope measurements in Nuvvuagittuq Greenstone Belt (Nunavik, Quebec). Proc. Natl. Acad. Sci. U. S. A. 112, 707–712.
- Thomazo, C., Ader, M., Farquhar, J., Philippot, P., 2009. Methanotrophs regulated atmospheric sulfur isotope anomalies during the Mesoarchean (Tumbiana Formation, Western Australia). Earth Planet. Sci. Lett. 279, 65–75.
- Thomazo, C., Nisbet, E.G., Grassineau, N.V., Peters, M., Strauss, H., 2013. Multiple sulfur and carbon isotope composition of sediments from the Belingwe Greenstone Belt (Zimbabwe): a biogenic methane regulation on mass independent fractionation of sulfur during the Neoarchean? Geochim. Cosmochim. Acta 121, 120–138.
- Turner, J.S., 1973. Buoyancy Effects in Fluids (Cambridge Monographs on Mechanics). Cambridge University Press, Cambridge.
- Ueno, Y., Ono, S., Rumble, D., Maruyama, S., 2008. Quadruple sulfur isotope analysis of ca. 3.5 Ga Dresser Formation: New evidence for microbial sulfate reduction in the early Archean. Geochim. Cosmochim. Acta 72, 5675–5691.
- Wacey, D., Noffke, N., Cliff, J., Barley, M.E., Farquhar, J., 2015. Micro-scale quadruple sulfur isotope analysis of pyrite from the similar to 3480 Ma Dresser Formation: New insights into sulfur cycling on the early Earth. Precambrian Res. 258, 24–35.
- Whitehill, A.R., Ono, S., 2012. Excitation band dependence of sulfur isotope massindependent fractionation during photochemistry of sulfur dioxide using broadband light sources. Geochim. Cosmochim. Acta 94, 238–253.
- Whitehill, A.R., Xie, C., Hu, X., Xie, D., Guo, H., Ono, S., 2013. Vibronic origin of sulfur mass-independent isotope effect in photoexcitation of SO₂ and the implications to the early earth's atmosphere. Proc. Natl. Acad. Sci. U. S. A. 110, 17697–17702.
- Whitehill, A.R., Jiang, B., Guo, H., Ono, S., 2015. SO₂ photolysis as a source for sulfur mass-independent isotope signatures in stratospehric aerosols. Atmos. Chem. Phys. 15, 1843–1864.
- Williford, K.H., Ushikubo, T., Lepot, K., Kitajima, K., Hallmann, C., Spicuzza, M.J., Kozdon, R., Eigenbrode, J.L., Summons, R.E., Valley, J.W., 2016. Carbon and sulfur isotopic signatures of ancient life and environment at the microbial scale: Neoarchean shales and carbonates. Geobiology 14, 105–128.
- Zerkle, A.L., Claire, M.W., Domagal-Goldman, S.D., Farquhar, J., Poulton, S.W., 2012. A bistable organic-rich atmosphere on the Neoarchaean Earth. Nat. Geosci. 5, 359–363.
- Zhelezinskaia, I., Kaufman, A.J., Farquhar, J., Cliff, J., 2014. Large sulfur isotope fractionations associated with Neoarchean microbial sulfate reduction. Science 346, 742–744.